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reaction. To begin, check that this equation is shown: $\text{Fe}_2\text{O}_3 + 3\text{CO} \rightarrow 2\text{Fe} + 3\text{CO}_2$. Look at the coefficients (such as the 3 in 3CO) in front of each substance in the equation.

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Since 20 moles NO_2 are produced, and it takes 3 moles O_2 to produce 4 moles of NO_2 , the answer would be 15 moles. $15 \text{ moles} = 20 \text{ moles } \text{NO}_2 \times (3 \text{ moles } \text{O}_2 / 4 \text{ moles } \text{NO}_2)$ --- This is the molar ratio between the two.

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