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Ascorbic acid is an organic compound with formula $\text{C}_6\text{H}_8\text{O}_6$, originally called hexuronic acid. It is a white solid, but impure samples can appear yellowish. It dissolves well in water to give

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mildly acidic solutions. It is a mild reducing agent.. Ascorbic acid exists as two enantiomers (mirror-image isomers), commonly denoted "l" (for "levo") and "d" (for "dextro").

~~Chemistry of ascorbic acid—Wikipedia~~

In organic chemistry, when a molecule with a planar ring obeys Hückel's rule, where the number of π electrons fit the formula $4n + 2$ (where n is an integer), it attains extra stability and symmetry. In benzene, the prototypical aromatic compound, there are 6 π bonding electrons ($n = 1$, $4n + 2 = 6$). These occupy three delocalized π molecular orbitals (molecular orbital theory) or form ...

~~Covalent bond—Wikipedia~~

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