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KEY 7.2 moles NO
2 Answer each of
the following
questions using
the equation
provided. BE

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SURE TO BALANCE
EACH EQUATION
BEFORE SOLVING
ANY PROBLEMS.

SHOW ALL WORK .



moles of NO will
react with _____

mole(s) of O 2

to produce _____

mole(s) of NO 2.

b. ? moles NO 2

= 3.6 moles O 2

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Worksheet 2
moles $0.2 = c.$

CHEMISTRY: A

Study of Matter

Mole Worksheet

(Dimensional

Analysis) #2 I.

What is the mass

(in grams) for

each of the

following

compounds or

elements? 1.

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Answer Key

7.24 moles of
silver phosphate
2. 2.88 moles of
diphosphorous
pentoxide 3.
0.0009273 moles
of zinc
bicarbonate 4.
154.8 moles of
silicon
tetraiodide 5.
88.624 moles of
silver II.

Answer the

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(Dimensional
Analysis) #2
mole practice
problems answer
key provides a
comprehensive
and
comprehensive
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discover many
creative ideas
from themselves.

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(Partial Key) 1.
A sample of

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sodium metal is determined to have a mass of 5.43 grams. How many moles of sodium are in this sample? 2. What is the molar mass of C_2H_6 ? 3.

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Molar Mass
Worksheet -

Answer Key

Calculate the
molar masses of
the following
chemicals: 1) Cl
2) 71 g/mol 2)
KOH 56 Mole
worksheet #2
answer key with
work. 1) 1 g/mol 3)
BeCl 2) 80 g/mol

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4) FeCl_3 162.3
g/mol 5) BF_3

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Mole Calculation
Worksheet -

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Worksheet 2 Answer Key 1)

How many moles
are in 15 grams
of lithium? 0.46

moles 2) How
many grams are
in 2.4 moles of
sulfur? 77.0

grams 3) How
many moles are
in 22 grams of
argon? 0.55

moles 4) How
many grams are

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Worksheet 2
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in 88.1 moles of
magnesium? 2141
grams 5) How
many moles are
in 2.3 grams of
phosphorus?
0.074 moles

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Stoichiometry
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Key Gram Formula
Mass Worksheet

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Worksheet. There
are three mole
equalities. They
are: 1 mol =
 6.02×10^{23}
particles. 1 mol
= g-formula-mass
(periodic table)
1 mol = 22.4 L
for a gas at
STP. Each
equality can be
written as a set

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of two conversion factors. They are: Mole-Particle Conversions. 1. How many moles of magnesium is 3.01×10^{22} atoms of magnesium? 3.01×10^{22} atoms = 5×10^{-2} moles. 2.

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The Results for

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is 384 403 km

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Key Calculate

the mass in

grams of each of

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6 answer key. 8.

00 moles of

aluminum 6. 7.

00 moles of

iodine (I 2) 4

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00 x 10² moles

of chlorine (Cl

2

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Balance this
equation: $\text{N}_2 +$
 $\text{H}_2 \rightarrow \text{NH}_3$,
write the

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Worksheet 2
following molar
ratios: N_2 / H_2
 N_2 / NH_3 $\text{H}_2 /$

NH_3 . 2) Balance
this equation:
 $\text{H}_2 + \text{S} \rightarrow \text{H}_2\text{S}$,

write the
following molar
ratios: $\text{H}_2 / \text{H}_2\text{S}$
 H_2 / S $\text{H}_2\text{S} / \text{S}$.

3) Write and
balance the
equation for the
synthesis of

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water. Then
answer the
following
questions.

Mole Ratio Worksheet

Chapter 2: Atoms
and Atomic Molar
Mass Worksheet
and Key 1.

Complete the
following table:
Symbol Symbol-

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Mass Number 2

Atomic # Mass #

protons

neutrons #

electrons ${}^3\text{Li}$ 7

Mo-96 49 53 72

35 ${}^{92}\text{U}$ 238

Answer the

following

questions. Be

sure to write

units with every

number and to

use the correct

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number of
significant
figures.
Answer Key

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