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In physics, a partition function describes the statistical properties of a system in thermodynamic equilibrium. [citation needed] Partition functions are functions of the thermodynamic state variables, such as the temperature and

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volume. Most of the aggregate thermodynamic variables of the system, such as the total energy ...

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study of the energy

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specific heat, latent heat,  
and enthalpy of a system

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Relation to heat and  
internal energy. In  
thermodynamics, heat is  
energy in transfer to or  
from a thermodynamic  
system, by mechanisms

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other than thermodynamic work or transfer of matter. Heat refers to a quantity transferred between systems, not to a property of any one system, or 'contained' within it. On the other hand, internal energy is a property of a single system.

~~Thermal energy—~~

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Thermodynamics and  
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comes from the Greek  
words ‘ therme ’ and  
‘ dynamic which means  
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by a presentation of the  
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