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A change in physical form of a substance

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or physical properties of a substance like color, melting point, state, size, shape, and arrangement of molecules is known as physical change. A change in chemical properties of a substance like composition, reactivity with water, corrosiveness, and pH is known as chemical change. In chemical change, a new substance will be

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SOLUTIONS MANUAL 35. Aqueous
solutions of potassium iodide and silver

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Solutions
nitrate are mixed, forming the precipitate
silver iodide. chemical equation: $KI(aq)$
 $AgNO_3(aq) \rightarrow KNO_3(aq) + AgI(s)$
complete ionic equation: $K^+(aq) + I^-(aq) + Ag^+(aq) + NO_3^-(aq) \rightarrow K^+(aq) + NO_3^-(aq) + AgI(s)$

Chemical Reactions
Chemical Reactions

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Solution. Physical: boiling and melting are physical changes. When water boils, no bonds are broken or formed. The change could be written: $\text{H}_2\text{O} (l) \rightarrow \text{H}_2\text{O} (g)$ Chemical: The dark grey nail changes color to form an orange flaky substance (the rust); this must be a

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chemical change.

3.6: Changes in Matter: Physical and
Chemical Changes ...

240 Chemistry: Matter and Change □

Chapter 12 Solutions Manual CHAPTER
12 SOLUTIONS MANUAL 30. Compare
and contrast sublimation and evaporation.

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In both processes, the substances become a vapor. During sublimation, the substance goes from the solid phase directly to the vapor phase. During evaporation, particles in a liquid gain enough

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